## Communications to the editor

## OGANOMYCINS, NEW 7-METHOXY-CEPHALOSPORINS PRODUCED BY PRECURSOR FERMENTATION WITH HETEROCYCLIC THIOLS

Sir:

In the search for novel methoxycephalosporins, we encountered new, 3-substituted 7-methoxycephalosporins when heterocyclic thiols (*cf.* RSH's in Table 1) were added to fermenting broth of an actinomycete, *Streptomyces oganonensis* n. sp. strain Y-G19Z, which was isolated from a soil sample collected at Ogano Town in Chichibu, Saitama Prefecture, Japan, and was shown by us to produce methoxycephalosporins.<sup>1)</sup> The new, methoxycephalosporins were named oganomycins.

Oganomycin G was produced as a main antibiotic when Y-G19Z strain was grown at 30°C in a medium adjusted to pH 7.0 which contained 7% starch, 2% gluten meal, 2% soybean meal, 0.8% glycerol, 0.1% Casamino acid, 0.01% ferric sulfate and 0.2% of a precursor (5-mercapto-1-methyl-1H-tetrazole). After  $4 \sim 5$  days when the potency of oganomycin G, assayed by the paper disk-plate method against Proteus mirabilis IMF-OM9, reached a maximum, the broth was harvested and filtered. Oganomycin G in the filtrate was adsorbed on Amberlite XAD-2 at pH 3.0 and eluted with 50% acetone. The eluate was concentrated under reduced pressure. The antibiotic in it was adsorbed on Amberlite IRA-68 (Cl<sup>-</sup>) and eluted with 1 M sodium nitrate solution at pH 7.2 containing 0.1 м sodium acetate. After repetition of XAD-2 adsorption and acetone elution as abovestated, the antibiotic was chromatographed on DEAE-Sephadex A-25 (CH<sub>3</sub>COO<sup>-</sup>) equilibrated and eluted with 0.5 M ammonium bromide solution (adjusted to pH 3.1 with acetic acid). Active fractions, after XAD-2 adsorption and acetone elution, were further chromatographed on a cellulose powder (Avicel) using successively three different solvent systems, *i.e.* isopropanol water (7:3), isopropanol - *n*-butanol - acetic acid - water (21:3:7:9) and *n*-butanol - acetic acid - water (4:1:2). Active fractions showing a single specific spot on an Avicel thin-layer

chromatogram were finally chromatographed on Sephadex G-10 with distilled water. Active fractions thus obtained were lyophilized to give white powder of oganomycin G which was found to show a single peak on a high performance liquid chromatogram.

Oganomycin G was obtained as an amphoteric, hygroscopic, colorless amorphous powder melting at  $160 \sim 170^{\circ}$ C with decomposition. It was easily soluble in water, sparingly soluble in methanol, but insoluble in chloroform, ether, ethyl acetate and *n*-hexane. It was stable at pH 1.5~8.0 in aqueous solution. The elemental analysis was as follows: calcd. for C<sub>17</sub>H<sub>22</sub>N<sub>7</sub>O<sub>7</sub>S<sub>2</sub>-Na·2H<sub>2</sub>O: C 36.49, H 4.68, N 17.52, S 11.46; found: C 37.48, H 4.25, N 16.74, S 10.90. It showed a maximum in UV spectrum at 273 nm in 0.01 M phosphate buffer at pH 6.4 (E<sup>1%</sup><sub>16m</sub> 204). The IR spectrum is presented in Fig. 1. The PMR chemical shifts are given in Table 1. It gave a positive ninhydrin reaction.

For further purification, oganomycin G was converted to glutaryl derivative (organomycin GG) by oxidative deamination at its  $\alpha$ -aminoadipyl moiety using D-amino acid oxidase from Trigonopsis variabilis IFO-07552). Oganomycin G was treated with Triton X 100-activated Trigonopsis variabilis cells at 33°C under aeration in a phosphate buffer at pH 8.1 containing 0.026% sodium azide and adequate amount of hydrogen peroxide. After the ninhydrin test changed to negative, the reaction mixture was centrifuged at 4°C. Oganomycin GG in the supernatant was extracted repeatedly with ethyl acetate at pH  $1.5 \sim 2.0$ . The extracted antibiotic was transferred to phosphate buffer at pH 6.0, retransferred to ethyl acetate: and, after removal of the solvent, chromatographed on Avicel with *n*-butanol - acetic acid - water mixture (4:1:2). Active fractions were lyophilized to give white powder of pure oganomycin GG, which gave needle crystals on recrystallization from methyl ethyl ketone.

Oganomycin GG was obtained as acidic, colorless needles melting at 163.5°C with decomposition. It was easily soluble in methanol and ethanol; soluble in water, isopropanol, *n*-butanol, benzyl alcohol, ethyl acetate, methyl ethyl ketone

		res and PMR chemical	sinits of oganomychis.	
R of precursor RSH	N−N L CH3	SCH2COOH	_N_N_CH3	_K <sup>N−N</sup>
Structure		HOOC H <sub>2</sub> N <sup>-CHCH</sup> 2CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CONF	OCH <sub>3</sub> S O CH <sub>2</sub> SR COOH	·
Chemical shift $(\hat{o}, ppm; D_2O)$	Oganomycin G	Oganomycin F	Oganomycin H	Oganomycin I
3', 4'-CH <sub>2</sub> CH <sub>2</sub>	2.36, 4H, m	2.35, 4H, m	2.34, 4H, m	2.30, 4H, m
$2'-CH_2$	2.96, 2H, m	2.96, 2H, m	2.95, 2H, m	2.93, 2H, m
5'-CH	4.38, 1H, m	4.25, 1H, m	4.26, 1H, m	4.26, 1H, m
2 –CH <sub>2</sub>	3.83~4.34, 2H dd, <i>J</i> =18 Hz	3.73~4.33, 2H dd, <i>J</i> =18 Hz	3.72~4.31, 2H dd, <i>J</i> =18 Hz	3.69~4.29, 2H dd, <i>J</i> =18 Hz
3 -CH <sub>2</sub>	4.47~4.83, 2H dd, <i>J</i> =14 Hz	4.42~4.91, 2H dd, <i>J</i> =14 Hz	4.40~4.95, 2H dd, <i>J</i> =14 Hz	4.45~4.99, 2H dd, <i>J</i> =14 Hz
6 –CH	5.59, 1H, s	5.63, 1H, s	5.63, 1H, s	5.56, 1H, s
7 –OCH <sub>3</sub>	3.98, 3H, s	4.00, 3H, s	3.99, 3H, s	3.97, 3H, s
Heterocyclic proton	4.50, 3H, s	4.44, 2H, s	3.19, 3H, s	9.85, 1H, s
Structure	HOOCCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CONH CH <sub>2</sub> SR			
Chemical shift $(\delta, \text{ ppm}; \text{CD}_3\text{OD})$	Oganomycin GG	Oganomycin GF	Oganomycin GH	Oganomycin GI
3'-CH2	1.94, 2H, m	1.93, 2H, m	1.92, 2H, m	1.92, 2H, m
2', 4'-CH <sub>2</sub> CH <sub>2</sub>	2.40, 4H, m	2.41, 4H, m	2.40, 4H, m	2.40, 4H, m
2 –CH <sub>2</sub>	3.40~3.83, 2H dd, <i>J</i> =18 Hz	3.37~3.81, 2H dd, <i>J</i> =18 Hz	3.38~3.80, 2H dd, <i>J</i> =18 Hz	3.39~3.85, 2H dd, <i>J</i> =18 Hz
3 –CH <sub>2</sub>	4.17~4.50, 2H dd, <i>J</i> =14 Hz	4.15~4.63, 2H dd, <i>J</i> =14 Hz	4.17~4.64, 2H dd, <i>J</i> =14 Hz	4.24~4.73, 2H dd, <i>J</i> =14 Hz
6 –CH	5.02, 1H, s	5.05, 1H, s	5.03, 1H, s	5.03, 1H, s
7 –OCH <sub>3</sub>	3.51, 3H, s	3.51, 3H, s	3.51, 3H, s	3.51, 3H, s
Heterocyclic proton	3.99, 3H, s	4.11, 2H, s	2.71, 3H, s	9.35, 1H, s

Table 1. Structures and PMR chemical shifts of oganomycins.

and methyl propyl ketone; but insoluble in chloroform, ether, benzene and *n*-hexane. It was stable at pH  $1.5 \sim 8.0$  in aqueous solution. It gave following analysis: calcd. for C<sub>16</sub>H<sub>20</sub>N<sub>6</sub>O<sub>7</sub>S<sub>2</sub>: C 40.67, H 4.27, N 17.79, S 13.57; found: C 40.80, H 4.27, N 17.59, S 13.29. It gave a UV absorption maximum at 269 nm in 0.01 M phos-

phate buffer at pH 6.5 ( $\varepsilon$  11,100). Its IR spectrum is given in Fig. 2. The PMR chemical shifts are shown in Table 1. It gave a negative ninhydrin reaction.

Oganomycin G gave  $\alpha$ -aminoadipic acid after 6 N hydrochloric acid hydrolysis, and 5-mercapto-1-methyl-1H-tetrazole (*m/e* 116), the precursor

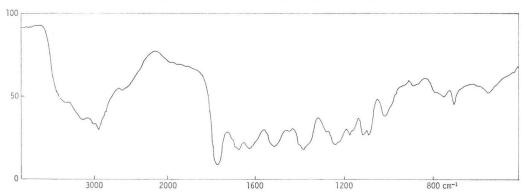
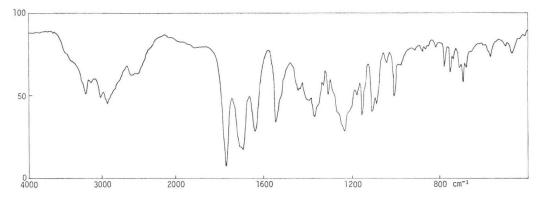


Fig. 1. IR spectrum of oganomycin G (KBr).





moiety, after Dowex 50W (H+) hydrolysis in methanol. From these results and those indicating the presence of 7-methoxy cephem skeleton (IR absorption at 1765 cm<sup>-1</sup> due to  $\beta$ lactam; PMR signals ( $\delta$ , ppm) in D<sub>2</sub>O at 3.83~ 4.34 (2H, dd, J=18 Hz), 5.59 (1H, s) and 3.98 (3H, s) and also from a PMR signal at 4.50 (3H, s) which indicated N-methyl protons of tetrazole moiety, the structure given in Table 1 was postulated for oganomycin G. Oganomycin GG gave glutaric acid (m/e 276 after trimethylsilylation) instead of  $\alpha$ -aminoadipic acid after hydrochloric acid hydrolysis. Further, from the IR absorption bands and the PMR signals presented in Table 1 and from the elemental analysis, the structure of oganomycin GG was assigned as given in Table 1. A <sup>13</sup>C-NMR study supported this structure. Details will be reported elsewhere.3)

By similar procedures, we obtained oganomycins F, H, I and GF, GH and GI as illustrated in Table 1. When disulfides, RSSR's, were used instead of RSH's in Table 1, the same 3-substituted derivatives were obtained.

Oganomycin F, oganomycin H and oganomycin I were obtained as amphoteric, hygroscopic, colorless amorphous powders showing solubilities and stabilities similar to those of oganomycin G and giving a positive ninhydrin reaction. They gave similar IR spectra, and liberated  $\alpha$ -aminoadipic acid and precursor moieties after hydrolysis. Their chemical shifts are given in Table 1. Oganomycin F showed: m.p. at 156~170°C (dec.);  $\lambda_{max}$  at 287 nm in phosphate buffer at pH 6.4 (E<sup>1%</sup><sub>1cm</sub> 240); analysis calcd. for C<sub>19</sub>H<sub>21</sub>N<sub>5</sub>O<sub>9</sub>S<sub>4</sub>Na<sub>2</sub>: C 35.79, H 3.32, N 10.98, S 20.11; found: C 35.95, H 3.87, N 10.85, S 18.33. Oganomycin H showed: m.p. at 170°C (dec.);  $\lambda_{\rm max}$  at 272 nm at pH 6.4; analysis calcd. for  $C_{18}H_{22}N_5O_7S_3Na \cdot 2H_2O$ : C 37.56, H 4.55, N 12.17, S 16.71; found: C 37.82, H 4.01, N 12.90, S 14.97. Oganomycin I showed: m.p. at 175~ 180°C (dec.);  $\lambda_{max}$  at 274 nm at pH 6.4 (E<sup>1%</sup><sub>1cm</sub> 232); analysis calcd. for  $C_{17}H_{21}N_5O_7S_3 \cdot 2H_2O$ :

	MIC (µg/ml)*		
Test organism	Oganomycin G	Oganomycin H	Oganomycin I
Bacillus megaterium 10778	6.25	12.5	12.5
Bacillus subtills ATCC 6633	6.25	12.5	6.25
Staphylococcus aureus Smith	100	50	100
Staphylococcus aureus Terashima	50	25	50
Staphylococcus aureus Onuma	50	50	100
Micrococcus flavus	1.56	3.13	1.56
Sarcina lutea	6.25	25	6.25
Corynebacterium xerosis	6.25	12.5	6.25
Mycobacterium 607	50	50	100
Mycobacterium phlei	50	50	100
Escherichia coli O–1	3.13	6.25	3.13
Escherichia coli NIHJ	3.13	6.25	6.25
Klebsiella pneumoniae ATCC 10031	3.13	3.13	6.25
Vibrio HY 133	6.25	12.5	6.25
Salmonella cholerae-suis 1348	1.56	3.13	1.56
Salmonella typhi H901W	3.13	3.13	3.13
Salmonella enteritidis 1891	1.56	1.56	1.56
Shigella flexneri 2a 1675	3.13	6.25	6.25
Shigella sonnei II 37148	3.13	6.25	6.25
Proteus vulgaris OXK US	1.56	3.13	3.13
Proteus mirabilis IFM–OM9	1.56	3.13	3.13
Pseudomonas aeruginosa ATCC 8689	100	50	100
Pseudomonas ovalis IAM 1002	100	50	100
Pseudomonas melanogenum	100	50	100
Pseudomonas aeruginosa 99 (GM–R)	100	50	100

Table 2. Antibacterial spectra of oganomycins.

\* Determined on heart infusion agar inoculated with one loopful of overnight culture.

MIC's of cephamycin C are reported as follows (in  $\mu$ g/ml): Staphylococcus aureus 800; Streptococcus agalactiae 400; Streptococcus pyogenes 200; Diplococcus pneumoniae 200; Escherichia coli 13 ~ 50; Klebsiella pneumoniae 6~25; Aerobacter aerogenes 100~>800; Salmonella pullorum 13; Salmonella schottmuelleri 25; Shigella 13; Proteus mirabilis 0.4~6; Proteus morganii 13; Proteus vulgaris 3~13; Pseudomonas aeruginosa >800. (Antimicr. Agents & Chemoth. 2: 281~286, 1972).

C 37.84, H 4.67, N 12.98, S 17.83; found: C 37.53, H 4.36, N 12.77, S 16.42.

Oganomycins GF, GH and GI were obtained as acidic, colorless amorphous powders showing solubilities and stabilities similar to oganomycin GG and giving negative ninhydrin reactions. They gave PMR signals presented in Table 1 and had IR spectra similar to oganomycin GG. They gave glutaric acid after hydrochloric acid hydrolysis. Oganomycin GF showed: m.p. at  $75 \sim 78^{\circ}$ C;  $\lambda_{max}$  at 278 nm at pH 6.5 ( $\varepsilon$  13,800); analysis calcd. for C<sub>18</sub>H<sub>20</sub>N<sub>4</sub>O<sub>9</sub>S<sub>4</sub>·2H<sub>2</sub>O: C 35.99, H 4.03, N 9.33; found: C 35.77, H 3.81, N 9.42. Oganomycin GH showed: m.p. at 95~99°C;  $\lambda_{max}$  at 273 nm at pH 6.5 ( $\varepsilon$  11,300); analysis calcd. for  $C_{17}H_{20}N_4O_7S_8$ : C 41.79, H 4.13, N 11.47; found: C 41.89, H 4.27, N 11.17. Oganomycin GI showed: m.p. at 88~92°C,  $\lambda_{max}$  at 274 nm at pH 6.5 ( $\varepsilon$  13,000); analysis calcd. for  $C_{16}H_{18}N_4O_7S_8 \cdot 1/2H_2O$ : C 39.74, H 3.96, N 11.59; found: C 39.69, H 3.87, N 11.32.

Oganomycins showed the following Rf's on silica gel thin-layer chromatogram developed with chloroform - methanol - formic acid mixture (80: 20: 2): G 0.02, F 0.00, H 0.02, I 0.02, GG 0.49, GF 0.35, GH 0.59, GI 0.49; on Avicel SF with *n*-butanol - acetic acid- water (4: 1: 2): G 0.34, F 0.37, H 0.44, I 0.36, GG 0.64, GF 0.72, GH 0.77 and GI 0.65. Oganomycins gave the following retention times on a high performance liquid chromatography using  $\mu$  Bondapak C<sub>18</sub> (Waters Co.) and solvent mixture of acetonitrile water (10: 90) adjusted to pH 3.3 with acetic acid (' and '' mean minute and second): G 1'53'', F 3'14'', H 2'55'', I 1'56'', GG 4'54'', GF 13'24'', GH 11'18'' and GI 5'28''.

Oganomycins showed antibacterial activities against Gram-positive and -negative bacteria. MIC's of oganomycins G, H and I are given in Table 2. Oganomycins GG, GH and GI were about 1/2 to 1/8 as active as their parent compounds.

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